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can electrons NOT
exist. in between

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orbits. what are atomic energy levels in the atom. the electron orbits. when an atom is in an excited state the electrons are in the _____ orbitals. farther.

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ARRANGEMENT OF
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Development ...**

The Arrangement of
Electrons in Atoms
chapter of this Holt
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Course helps students learn the essential lessons associated with how electrons are arranged in atoms.

**Holt McDougal
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Chapter 4:
Arrangement of ...**

1s 2s 2p 3s 3p

3d 4s 1 PROBLEMS

Write the answer on the line to the left.

Show all your work in the space provided. 9.

1. 10^{12} m What is the

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wavelength of light
that has a frequency of
 3×10^4 Hz in a vacuum?
10. 3.3×10^{19} J What is
the energy of a photon
that has a frequency of
 5.0×10^{14} Hz? 32

ARRANGEMENT OF
ELECTRONS IN ATOMS
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**4 Arrangement of
Electrons in Atoms**

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Chapter Test Name _____

Class Date Chapter _____

Test A, continued _____

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4. The electron configuration below violates a. the Pauli exclusion principle. b. the Aufbau principle. c. Hund's rule. d. Both (a) and (c) _____
5. A photon is emitted from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell. b ...

**Assessment Chapter
Test A**

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4 - Reactions in

Aqueous Solution: Part

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In this video, I'll teach

you the meaning of the

terms solvent, solute,

electrolyte, and

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Test A, continued Use

this figure to answer
questions 7 and 8.

_____ 7. A solution
containing 35 g of Li
2SO 4 dissolved in 100
g of water is heated
from 10°C to 90°C.

According to
information in the
figure, this
temperature change
would result in a. an

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additional 5 g of Li₂SO₄ in ...

**Assessment Chapter
Test A - Ed W. Clark
High School**

Modern Chemistry 1
Arrangement of
Electrons in Atoms
CHAPTER 4 REVIEW
Arrangement of
Electrons in Atoms
Teacher Notes and
Answers Chapter 4
SECTION 1 SHORT
ANSWER 1. In order for
an electron to be

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ejected from a metal surface, the electron must be struck by a single photon with at least the minimum energy needed to knock the electron loose.

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